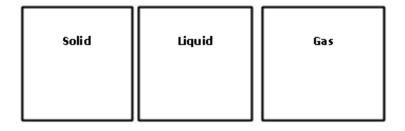
#### **States of matter**

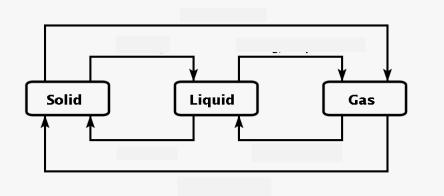
Draw the particle arrangement in each state of matter



#### 1. Solid, liquid or gas?

- I can be compressed \_\_\_\_\_\_
- I can hold my shape \_\_\_\_\_\_
- I have a fixed volume (size) \_\_\_\_\_ and \_\_\_\_\_
- I can expand to fill a container \_\_\_\_\_
- My particles can only vibrate \_\_\_\_\_\_\_
- 2. On this diagram below, name the 6 **changes in state** by writing the words above the correct arrows.

Using a red and a blue pencil, go over the arrows. Use red for when the temperature is going up and blue for when the temperature is going down



#### **Acids and Alkalis**

#### 3. Concentrated or dilute?

Copy the sentences below choosing the correct bold words:

- 1. Acids taste sour/feel soapy
- 2. Some acidic and alkaline solutions are corrosive/correlated
- **3.** A concentrated solution of an acid is **more/less** corrosive than a dilute solution



	рН 1	рН 2	рН 3	рН 4	рН 5	рН 6	рН 7	рН 8	рН 9	pH 10	рН 11	pH 12	рН 13	рН14	0000
--	------	------	------	------	------	------	------	------	------	-------	-------	-------	-------	------	------

### 4. The pH scale

- 1. Colour in the pH scale for a strong acid, weak acid, neutral, weak alkali and a strong alkali
- 2. Give an example of a strong acid -
- 3. Give an example of a strong alkali -

#### 5. Complete the chemical equations:

hydrochloric acid + sodium hydroxide -> chloride + w

hydrochloric acid + calcium carbonate -> \_\_\_\_\_ chloride + w\_\_\_\_ + c\_\_\_\_ d\_\_\_

#### 1. Match up the Hazard symbol



Health Irritant/ hazard harmful	Corrosive	Damages the environment	Toxic
------------------------------------	-----------	-------------------------	-------

2. What is an **independent** variable?

Where should this variable go on a table of results?

3. What is a **dependent** variable?

Where should this variable go on a table of results?

4. What are control variables?

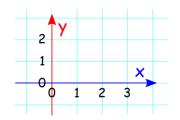
Why is it important to only change one thing?

5. How can you check if something is **reliable**?

6. Which axis?

Independent variable on \_\_\_\_ axis.

Dependent variable on \_\_\_\_ axis



#### Science Skills and Safety

- 6. Examples of a **continuous** variable are:
- 1)
- 2)
- 3)

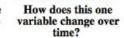
Which type of graph would you plot if **both variables are continuous**?

- 7. Examples of a **discontinuous** variable are.
- 1)
- 3)
- 3)

Which type of graph would you plot if **one variable is discontinuous?** 

#### BAR GRAPHS

How different are these variables to each other?



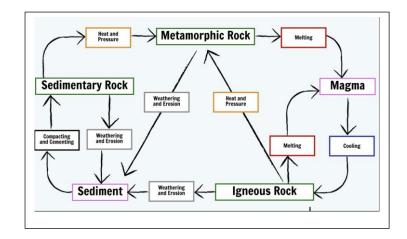


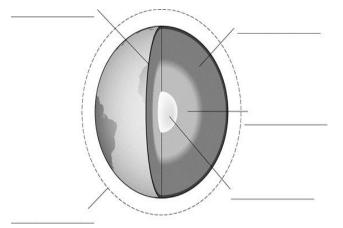


LINE GRAPHS

## The Rock Cycle

## **Layers of the Earth**





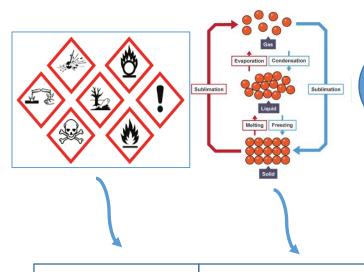
Earth's c	rust is b	oroken into
	and	that float on I move due to nts.
	•	s ago these plates continent called
7% <u>-</u>		

Igneous Rock
Formed when molten rock ( I/m cools so that it becomes s
Igneous rocks contain lots of c
Extrusive igneous rock is formed when
Intrusive igneous rock is formed when

	_
Describe how Metamorphic Rocks are formed	Formation of
	Rocks
	E1244074E
	1 14 14 17 15 15 15 15 15 15 15 15 15 15 15 15 15



Sedimentary Rocks
T
Bits of s are carried away by w or w
Sedimentation
Eroded s end up in the water and begin to settle.
C
Layers build up and press down upon the lower ones.
Cementation
S crystals g the layers together. S R is formed
Examples of sedimentary rocks are:-
1)
2)
3)

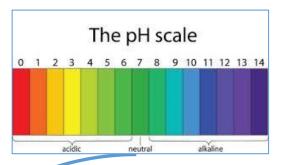


# What I know from primary science:

Particle patterns in solids liquids and gases plus changes of state in the water cycle

How to identify variables to change, measure and control in an investigation

How to measure temperature in degrees



General laboratory safety

Recognise hazard warning symbols and carry out basic risk assessments Know the properties of the different states of matter in terms of the particle model.

Describe changes of state in terms of the particle model.

Apply scientific terminology to investigations; correctly select either a bar chart or line graph to represent findings.

Define acids, alkali and neutralisation.

Use indicators to distinguish between an acid and an alkali.

Use the pH scale to measure acidity/alkalinity.

Know the composition and structure of the Earth.

Use the Rock Cycle to explain the formation of igneous, sedimentary and metamorphic rocks

#### **Future learning**

Concentration

Reaction of acids and bases with neutralisers

Extraction of metals from ores

Vocabulary: corrosive, indicator, neutral, pH scale, crust, mantle, core, tectonic plate, magma, laya, sediment, erosion, compaction, cementation, extrusive, intrusive, igneous, sedimentary, metamorphic, independent, dependent, control, reliable, continuous, discontinuous

